#### **CHEMICAL SYNTHESIS OF CERAMIC POWDER (PART 3)**

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#### Vapour Phase Reaction

Chemical Vapour Deposition (CVD)

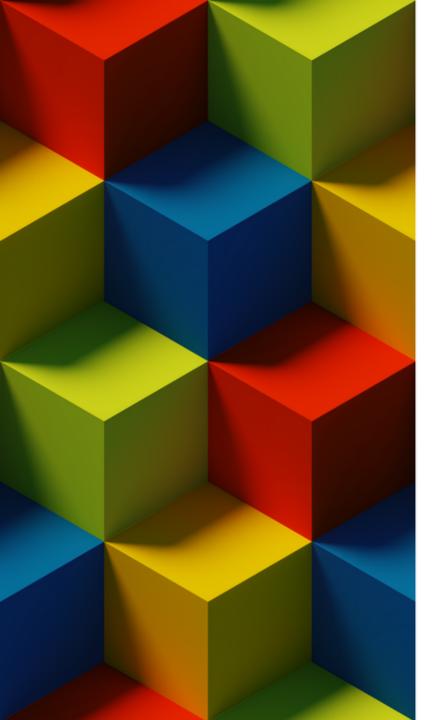
Thermal Deposition

DC Arc Plasma



#### Chemical Vapour Deposition (CVD)

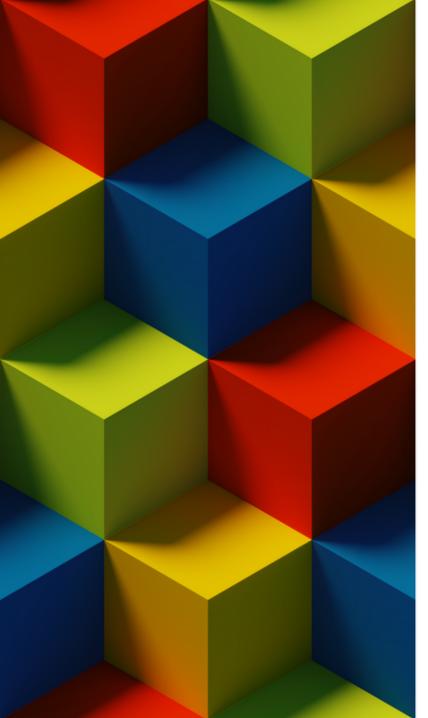
The formation of soot due to incomplete oxidation of firewood is probably the oldest example of deposition using CVD



## Chemical Vapour Deposition (CVD)

By definition...

CVD involves the dissociation and/or chemical reactions of gaseous reactants in an activated (heat, light, plasma) environment, followed by the formation of a stable solid product



#### What is CVD?

Solid material is obtained as a coating, a powder, or as single crystals

Example for Solid-Vapor Reaction

## Chemical Vapour Deposition (CVD)

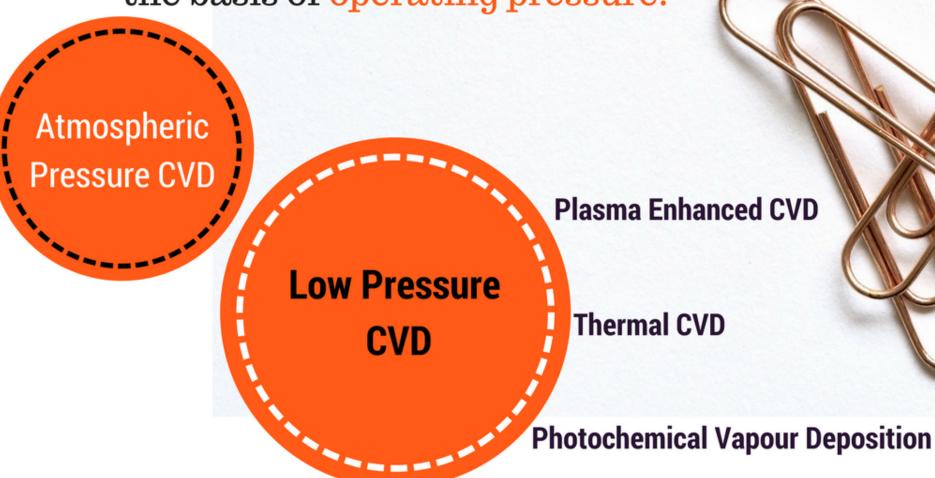
Chemical process used to produce:
high purity
high performance solid materials

Often used in semiconductor industries to produce thin film



#### Types of CVD

CVD's are classified into 2 types on the basis of operating pressure.



#### **Basic Concept**

Production of powder particle through:
Homogeneous or heterogeneous nucleation on
supplied nuclei.

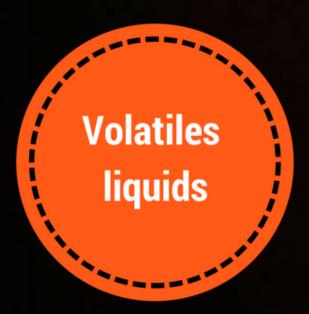
\*Nuclei : point where the crystal start to growth during solidification.

Reaction typically carried out by passing the reactants into a chamber (often through a burner tube with a gas flame).

#### **Basic Concept**

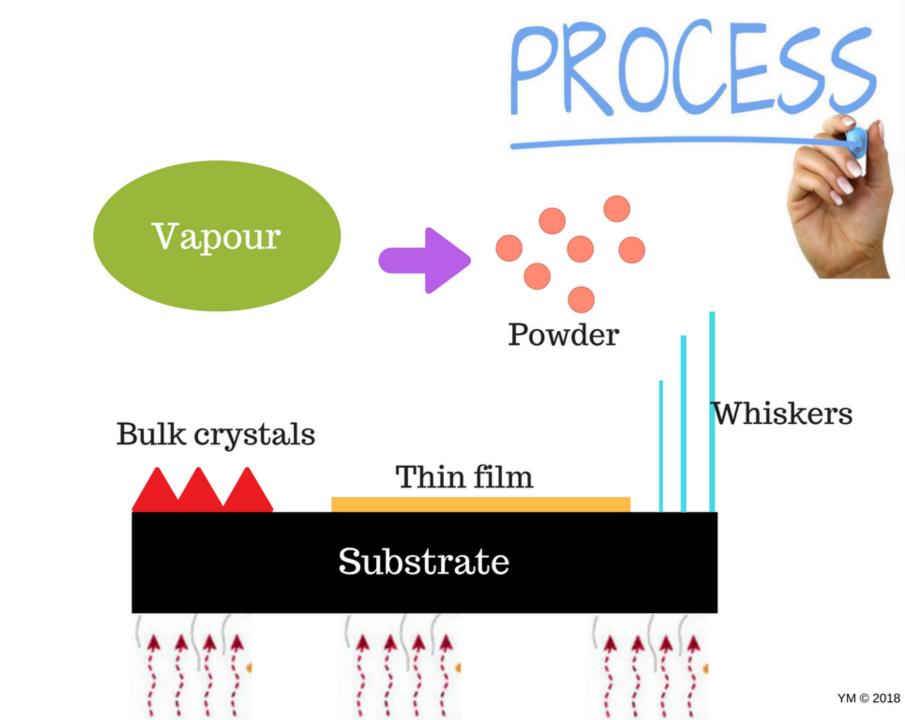
Reactants for vapour phase reaction...





Wide variety of alternative heat source such as plasmas and lasers.





Wafer (substrate)
exposed to one or
more volatile
reactants
(precursors)

Selected reactant materials is diffused in a \*carrier gas that flow over the hot substrate surface

PROCESS

heat from the hot substrate surface prompt chemical reactions between the reactant and carrier gas

volatile by-products are also produced- removed by gas flow through the reaction chamber

Desired deposit produced



1.Transport of reactants by forced convection to the deposition region

2.Transport of reactants by diffusion from the main gas stream to the substrate surface

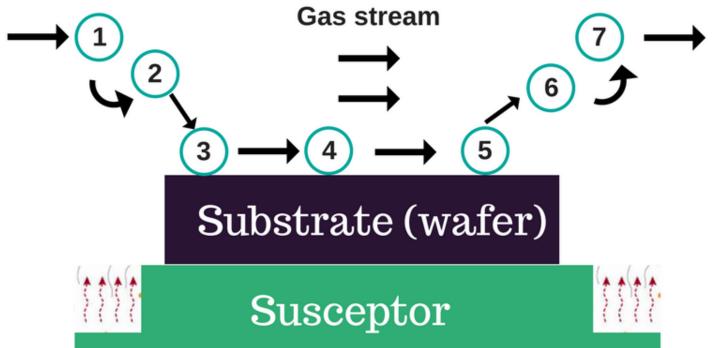
3.Adsorption of reactants on the wafer (substrate) surface

4.Chemical decomposition and other surface reactions take place

7.Transport of byproducts by forced convection away from the deposition region

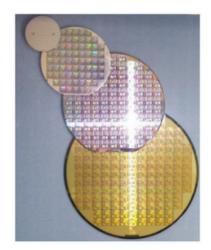
6.Transport of by-products by diffusion 5.Desorption of by-products from the surface

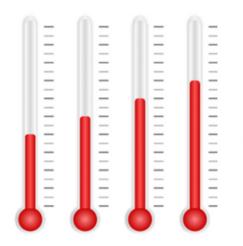




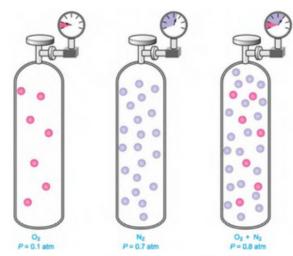
### Materials with different properties can be grown by varying the experimental conditions:

Substrate material





Substrate temperature



composition of the reaction gas mixture



total pressure gas flows



Versatile – can deposit any element or compound

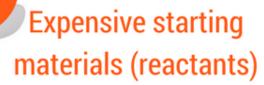
High Purity (~99.99%)

High Density - (~100% of theoretical)

Material formation well below Tm

Coating deposited conformal and near net shape

**Economical in production** 



Easily oxidized
(particularly when
producing non oxide
powder)



#### SiO<sub>2</sub>

$$SiH_4 + O_2 \rightarrow SiO_2 + 2H_2$$
 at  $400 - 500^{\circ}C$ 

#### $Si_3N_4$

$$3SiH_4 + 4NH_3 \rightarrow Si_3N_4 + 12H_2 \text{ at } 700 - 900^{\circ} \text{ C}$$

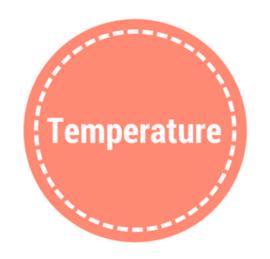
$$3SiCl_4 + 4NH_3 \rightarrow Si_3N_4 + 12HCl$$
 at  $850^{\circ}$  C

$$3SiH_2Cl_2 + 4NH_3 \rightarrow Si_3N_4 + 6HCl + 6H_2 \text{ at } 650 - 750^{\circ}C$$



Distance in the direction of gas flow



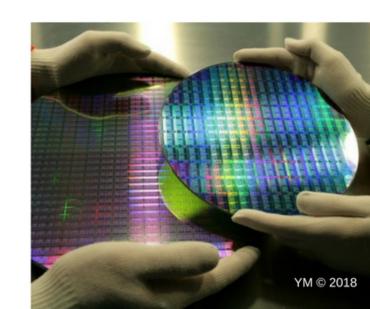


Velocity of carrier gas

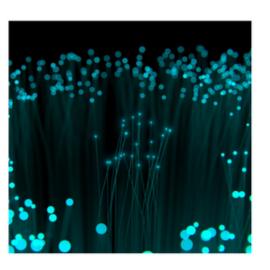
Better performance can be achieved by either increasing T and gas flow or by reducing pressure and gas velocity

However, increase in T adverse effect on the substrate due to incompatible CTE of the deposit and substrate

Reducing gas velocity nonuniformity of the deposition



# Applications of CVD



Optical Fibres for telecommunication



Synthetic diamond



Novel powders production





Gallium arsenide is used in some photovoltaic devices



Integrated circuits, sensors and optoelectronic devices.



Coatings for wear resistance, corrosion resistance, high temperature protection, erosion protection and combinations thereof